Colloidal quantum dots on macroscale perovskite single crystal

with perfect lattice matching

 Yu-Hao Deng^{1,2,3‡}, Yun-Gang Sang^{1,2,4‡}, Xiao-Wei Zhang^{1,2,5‡}, Yi-Fei Mao^{1,2,6}, Ren-Min Ma^{1,2*}
¹ State Key Lab for Mesoscopic Physics and School of Physics, Peking University, 100871 Beijing, China
² Frontiers Science Center for Nano-optoelectronics & Collaborative Innovation Center of Quantum Matter, Peking University, 100871 Beijing, China
³ Physics and Chemistry of Nanostructures Group, Ghent University, 9000 Ghent, Belgium
⁴ Institute of Microelectronics, Chinese Academy of Sciences, 100029 Beijing, China
⁵ Institute of Advanced Materials, Beijing Normal University, 100875 Beijing, China
⁶ School of Sensing Science and Engineering, School of Electronic Information and Electrical Engineering, Shanghai Jiao Tong University, Shanghai 200240, China
[‡] These authors contributed equally to this work
* Correspondence should be addressed to renminma@pku.edu.cn



Fig. S1. Synthesis process of the quantum-dot/perovskite heterocrystals. (a) The quantum dots with organic ligands were synthesized by colloidal method. Then the ligand organic-ligand molecules on the surface of quantum dots were exchanged by bromine ions. **(b)** Solution was dispensed on the perovskite substrate. **(c)** Quantum dots oriented attachment onto perovskite substrate, which along the same orientation direction between quantum dots and perovskite substrate. **(d)** Finally, perovskite epitaxial layer was formed by putting the quantum-dot/perovskite heterocrystals semiconductor into perovskite precursor solution under 80 °C.



Fig.S2 | (**a**) Absorption spectra of Perovskite and PbS quantum-dot/perovskite heterocrystals. (**b**) Photoluminescence imaging of the PbS quantum-dot/perovskite heterocrystals collected with a near infrared (top) and visible (bottom) camera. (**c**) Photoluminescence intensity of bare quantum dots and quantum-dot/perovskite heterocrystals under different quantum dots concentrations.